## SHORT STRUCTURAL PAPERS

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# New Investigation of the Structure of $\mathbf{M n}_{\mathbf{4}} \mathbf{A 1} \mathbf{1 1}_{1}$ 

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#### Abstract

Mn}_{4} \mathrm{Al}_{11}\), triclinic, $P \overline{1}, a=5.095$ (4), $b=$ 8.879 (8), $c=5.051$ (4) $\AA, \alpha=89.35$ (7), $\beta=$ $100.47(5), \gamma=105.08(6)^{\circ}, Z=1, V=216.8 \AA^{3}$, $D_{c}=3.96 \mathrm{Mg} \mathrm{m}^{-3}$, calculated from the idealized composition, and $3.88 \mathrm{Mg} \mathrm{m}^{-3}$, if the population factors are taken into account, $R(F)=0.067$ for 1180 reflections refined. This unit cell does not agree with the triclinic cell of Bland, but transforms properly to the monoclinic cell of the same author [Bland (1958). Acta Cryst. 11, 236-244]. The structure contains approximate layers consisting of Mn and Al atoms with a spacing of about $2 \cdot 2 \AA$. The short $\mathrm{Mn}-\mathrm{Al}$ and $\mathrm{Al}-\mathrm{Al}$ bonds given by Bland have been reconfirmed in this three-dimensional study.


Introduction. The structure of the alloy $\mathrm{Mn}_{4} \mathrm{Al}_{11}(\delta$ MnAl ) was first determined by Bland (1958) using four zones of reflections. The structure shows strong regularities in the coordination of the two nonequivalent Mn atoms, with two $\mathrm{Mn}-\mathrm{Al}$ bonds being much shorter than commonly occurring values (Bland, 1958). The structure can be approximately described as consisting of a series of layers parallel to (010), with an interplanar spacing of about $2 \cdot 2 \AA$. As a preliminary to an electron density study of the metal-metal bonding in this alloy, its structure was redetermined from threedimensional counter data.
Small single crystals were obtained from a lump of the material prepared at Argonne National Laboratory using a d.c. arc furnace. After cooling, the sample was sealed in an evacuated quartz tube and annealed for six days at 873 K . Unlike the preparation from a ternary melt described by Bland (1958), the method excludes

[^0]the possibility of Zn contamination. An irregular plate of approximate dimensions $0.2 \times 0.15 \times 0.06 \mathrm{~mm}$ was selected for data collection with Nb -filtered Mo $K \bar{\alpha}$ radiation ( $\lambda=0.71073 \AA$ ). Cell dimensions were obtained by least-squares refinement of the setting angles of 28 reflections with $2 \theta>34^{\circ}$. A full sphere of reflections ( $\sin \theta / \lambda<0.7 \AA^{-1}$ ) was measured at room temperature by an $\omega$ scan from $\omega-3^{\circ}$ to $\omega+1 \cdot 6^{\circ}$, the $\omega$ scan being selected because of the relatively large mosaic spread of the crystal.

2607 reflections were averaged after correction for Lorentz-polarization and absorption effects ( $\mu=$ $6.515 \mathrm{~mm}^{-1}$ ), to give 1254 unique reflections. The internal agreement factor between symmetry-related reflections was $R=0.033$. Reflections with $F<0$ were excluded yielding 1180 reflections for full-matrix leastsquares refinement. Positional and anisotropic thermal parameters were refined starting from Bland's (1958) atomic coordinates. Refinement of the occupancy of the positions led to a slight lowering of the $R$ factors which were $R(F)=0.071$ and 0.067 and $R_{w}(F)=$ 0.074 and 0.068 for the refinements without and with

Table 1. Positional parameters in the triclinic unit cell

|  | $x$ | $y$ | $z$ | Occupancy* |
| :--- | :--- | :--- | :--- | :--- |
|  | * |  |  |  |
| $\mathrm{Mn}(1)$ | $0.38915(19)$ | $0.13307(11)$ | $0.33365(17)$ | $1.000(6)$ |
| $\mathrm{Mn}(2)$ | $0.85543(22)$ | $0.39943(13)$ | $0.69743(22)$ | $0.924(6)$ |
| $\mathrm{Al}(0)$ | 0.0 | 0.0 | 0.0 | $0.489(7)$ |
| $\mathrm{Al}(1)$ | $0.5327(4)$ | $0.1255(2)$ | $0.8439(4)$ | $0.999(10)$ |
| $\mathrm{Al}(2)$ | $0.8932(4)$ | $0.1242(2)$ | $0.4833(4)$ | $0.991(10)$ |
| $\mathrm{Al}(3)$ | $0.3316(4)$ | $0.3717(2)$ | $0.5738(4)$ | $1.012(12)$ |
| $\mathrm{Al}(4)$ | $0.7196(5)$ | $0.3702(3)$ | $0.1802(5)$ | $0.997(10)$ |
| $\mathrm{Al}(5)$ | $0.1705(5)$ | $0.3214(3)$ | $0.0547(4)$ | $0.993(11)$ |

*Rescaled to make the occupancy of $\mathrm{Mn}(1)$ equal to one.
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$\mathrm{Mn}_{4} \mathrm{Al}_{11}$

Table 2. Mn-Al bond lengths $(\AA)$

|  | $\begin{gathered} \text { Bland } \\ (1958) \\ \sigma \sim 0.012 A \end{gathered}$ | $\begin{gathered} \text { Present } \\ \text { Study } \\ \sigma \sim 0.003 \end{gathered}$ |  | $\begin{gathered} \text { Bland } \\ (1958) \\ \sigma \sim 0.012 \mathrm{~A} \end{gathered}$ | $\begin{gathered} \text { Present } \\ \text { study } \\ \sigma \sim 0.003 \AA \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Mn}(1)-\mathrm{Al}(0)$ | 2.42 | 2.404 | $\mathrm{Mn}(2)-\mathrm{Al}(5)$ | 2.45 | 2.412 |
| $\mathrm{Al}(5)$ | $2 \cdot 50$ | 2.521 | $\mathrm{Al}(4)$ | 2.48 | 2.547 |
| $\mathrm{Al}(4)$ | 2.53 | 2.542 | $\mathrm{Al}(3)$ | 2.54 | 2.573 |
| $\mathrm{Al}(1)$ | $2 \cdot 54$ | $2 \cdot 553$ | $\mathrm{Al}(4)$ | 2.56 | $2 \cdot 576$ |
| $\mathrm{Al}(3)$ | $2 \cdot 53$ | 2.560 | $\mathrm{Al}(4)$ | 2.66 | 2.645 |
| $\mathrm{Al}(2)$ | $2 \cdot 54$ | 2.565 | $\mathrm{Al}(3)$ | 2.74 | 2.685 |
| $\mathrm{Al}(2)$ | $2 \cdot 61$ | $2 \cdot 622$ | $\mathrm{Al}(3)$ | 2.78 | 2.717 |
| $\mathrm{Al}(1)$ | 2.60 | 2.625 | $\mathrm{Al}(1)$ | 2.75 | 2.736 |
| $\mathrm{Al}(1)$ | 2.72 | 2.710 | $\mathrm{Al}(2)$ | 2.74 | 2.751 |
| $\mathrm{Al}(2)$ | 2.78 | 2.747 | $\mathrm{Al}(5)$ | 2.82 | 2.835 |
| Mn(2) | 3.20 | $3 \cdot 194$ | Mn(2) | 3.05 | 2.992 |
| $\mathrm{Mn}(1)$ | $3 \cdot 19$ | $3 \cdot 210$ |  |  |  |

the occupancy parameters respectively $\left\{R(F)=\sum \mid F_{o}\right.$ $-\left|F_{c}\right|\left|\sum\right| F_{o} \mid ; R_{w}(F)=\left(\sum w|\Delta F|^{2} / \sum w F_{o}^{2}\right)^{1 / 2}, w=$ $\left.1 / \sigma^{2}, \sigma(F)=\left[\sigma\left(F^{2}\right)^{2}+\left(0.03 F^{2}\right)^{2}\right]^{1 / 2}\right\}$. Extinction effects appeared to be insignificant. Final parameters are listed in Table 1,* while bond lengths are compared with earlier values in Table 2.

Discussion. The structural parameters of the earlier study are generally confirmed, though in some cases [such as $\mathrm{Mn}(2)-\mathrm{Al}(4)$ ] differences in bond lengths as large as $0.07 \AA$ are observed. The separation between the anomalously short and the one but shortest $\mathrm{Mn}-\mathrm{Al}$ bond for each of the coordination spheres is even more pronounced in the new results, the differences in length being 0.117 and $0.135 \AA$ for $\mathrm{Mn}(1)$ and $\mathrm{Mn}(2)$ respectively, rather than 0.08 and $0.03 \AA$ as observed previously (Table 2). Bond lengths of Bland (1958) in Table 2 are reproduced from Bland's triclinic parameters, and the triclinic cell obtained by transformation of his monoclinic cell. When the previously published triclinic cell is used, erroneous bond lengths are obtained, indicating a probable printing or copying error in the $\alpha$ value in the earlier publication.

The 'normal' value for the $\mathrm{Mn}-\mathrm{Al}$ bond length suggested by Nicol (1953) is $2.68 \AA$. Thus the $\mathrm{Mn}(1)-$ $\mathrm{Al}(0)$ bond of 2.404 (3) $\AA$, which is between layers, and the $\mathrm{Mn}(2)-\mathrm{Al}(5)$ bond of 2.412 (3) $\AA$ within the layer are confirmed to be unusually short. Each of the Mn atoms is surrounded by ten Al atoms which form a bicapped distorted cubic arrangement.

The thermal motion of both Mn atoms is quite anisotropic, the largest principal component of the thermalmotion tensors being about ten times larger than the smallest eigenvalues. The maximum amplitudes for both atoms are close to being perpendicular to the

[^1]

Fig. 1. The ten Al atoms surrounding (a) $\mathrm{Mn}(1)$ and (b) $\mathrm{Mn}(2)$. The $b$ axis is approximately vertical. The ellipsoids correspond to $75 \%$ probability.
layers, rather than to the short $\mathrm{Mn}-\mathrm{Al}$ bonds. The thermal motion is larger at the $\mathrm{Mn}(2)$ site, which also shows a deficiency in occupancy of about $8 \%$ according to the results of the least-squares refinement (Table 1). All other occupancies are not significantly different from the idealized values. Fig. 1 shows the vibration ellipsoids.

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## References

Bland, J. A. (1958). Acta Cryst. 11, 236-244.
Nicol, A. D. I. (1953). Acta Cryst. 6, 285-293.


[^0]:    * On leave from the Department of Physics, University of Helsinki, Helsinki, Finland.

[^1]:    * Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34847 ( 9 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

